

Atomic Mass of Carbon = 12

Atomic Mass of Hydrogen = 1

% Composition of Carbon = 80%

% Composition of Hydrogen = 20%

$$C = \frac{80}{12} = 6.67$$

$$H = \frac{20}{1} = 20$$

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Divide with the Smaller value

$$\begin{array}{r} C \\ 6.67 \\ \hline \end{array}$$

$$\begin{array}{r} 6.67 \\ \hline \end{array}$$

$$= 1$$

$$\begin{array}{r} H \\ 20 \\ \hline \end{array}$$

$$\begin{array}{r} 6.67 \\ \hline \end{array}$$

$$= 3$$

Empirical Formula =  $CH_3$

# 14 Waec Solutions To Essay And Objectives Chemistry Questions

**Robert R. Redfield**



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