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Chemistry Paper 3A WASSCE (SC), 2022

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Question 1A

A is an aqueous solution of iodine. **B** is 0.100 mol dm⁻³ sodium trioxothiosulphate (II).

- a. Put **B** into the burette. Pipette 20.0 cm³ or 25.0 cm³ of **A** into a conical flask. Add **E** from the burette until the reddish-brown colour fades to pale yellow, then add a few drops of starch indicator to obtain a dark blue solution. Continue adding **B** slowly from the burette until one drop of **B** causes the blue colour to disappear, leaving a colourless solution.

Repeat the titration to obtain concordant titre values.

Tabulate your results and calculate the average volume of **B** used.

The equation for the reaction is:



- b. From your results and the information provided, calculate the:

- i. Concentration in mol dm⁻³ of iodine in **A**;
- ii. mass in grammes of iodine in 1.00 dm³ of **A**.

[I=127.0]

[17 marks]

Observation

Majority of the candidates responded to this question and their performance was above average.

In part (a), majority of the candidates carried out the titration experiment correctly;

In part (b), majority of the candidates calculated the concentration in mol dm⁻³ of iodine in **A** and mass of iodine in 1.00 dm³ of **A**.

The expected answers include:

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**Andrew M. Olney,Irene-Angelica
Chounta,Zitao Liu,Olga C. Santos,Ig
lbert Bittencourt**

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Artificial Intelligence in Education Andrew M. Olney,Irene-Angelica Chounta,Zitao Liu,Olga C. Santos,Ig Ibert Bittencourt,2024-07-01 This book constitutes the refereed proceedings of the 25th International Conference on Artificial Intelligence in Education AIED 2024 held in Recife Brazil in July 8 12 2024 Proceedings The 49 full papers and 27 short papers presented in this book were carefully reviewed and selected from 334 submissions The papers present result in high quality research on intelligent systems and the cognitive sciences for the improvement and advancement of education

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
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