

2.1

CHEMICAL REACTIONS

Section Review

Objectives

1. Measure enthalpy using a calorimeter.
2. Identify and predict chemical reactions using phase diagrams.
3. Apply the law of conservation of energy to chemical reactions.

Vocabulary

- | | | |
|--------------------|---------------|----------------------------------|
| 1. endothermic | 2. exothermic | 3. enthalpy |
| 4. enthalpy change | 5. product | 6. law of conservation of energy |

Part A Completion

Use the information available to identify a chemical reaction, calculate, and write energy changes associated with the reaction using appropriate units and signs, identifying the system.

- | | |
|---|-----------|
| 1. Enthalpy change denotes either a heating or _____. | 6. _____ |
| 2. Enthalpy change is the _____ of a substance. | 7. _____ |
| 3. Enthalpy change is a _____ change. It is measured _____. | 8. _____ |
| 4. Enthalpy change is _____. | 9. _____ |
| 5. Enthalpy change is the _____ of a substance. | 10. _____ |
| 6. Enthalpy change is the _____ of a substance. | 11. _____ |
| 7. Enthalpy change is the _____ of a substance. | 12. _____ |
| 8. Enthalpy change is the _____ of a substance. | 13. _____ |
| 9. Enthalpy change is the _____ of a substance. | 14. _____ |
| 10. Enthalpy change is the _____ of a substance. | 15. _____ |
| 11. Enthalpy change is the _____ of a substance. | 16. _____ |
| 12. Enthalpy change is the _____ of a substance. | 17. _____ |
| 13. Enthalpy change is the _____ of a substance. | 18. _____ |
| 14. Enthalpy change is the _____ of a substance. | 19. _____ |
| 15. Enthalpy change is the _____ of a substance. | 20. _____ |

Part B True-False

Indicate whether each statement is true or false. If false, correct the statement.

1. Enthalpy change is the _____ of a substance.
2. Enthalpy change is the _____ of a substance.
3. Enthalpy change is the _____ of a substance.

4 Chemical Reactions Section Review Answer Key

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