

7. One gram of steam at 100°C condenses, and the water cools to 22°C .

a. How much heat is released when the steam condenses? 540

b. How much heat is released when the water cools from 100°C to 22°C ?

78 cal ($100 - 22 \times 1 \text{ cal/g}^{\circ}\text{C}$)

c. How much heat is released altogether? 618 cal



8. In a household radiator 1000 g of steam at 100°C condenses, and the water cools to 90°C .

a. How much heat is released when the steam condenses?

540,000 cal

b. How much heat is released when the water cools from 100°C to 90°C ?

10,000 cal

c. How much heat is released altogether?

550,000 cal



9. Radioactive minerals in common granite release 0.01 cal/kg of energy per year. If a 1-kg chunk of 50°C granite is thermally insulated, so all this energy heats it, how many years does it take to reach its melting temperature of 700°C ? (Assume the specific heat of granite is $200 \text{ cal/kg}^{\circ}\text{C}$.)

$Q = mc\Delta T = 1 \text{ kg} (200 \text{ cal/kg}^{\circ}\text{C}) (650^{\circ}\text{C}) = 130,000 \text{ cal} / 0.10 \text{ cal/year} = 1,300,000 \text{ yr}$

10. How many years would be required if the chunk of granite had a mass of 1 million kg? Why?

SAME! (MORE ROCK = MORE RADIATION)

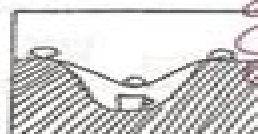
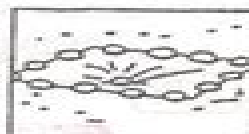
11. To calculate the time it takes to melt the 700°C granite, what other information would you need?

HEAT OF FUSION FOR GRANITE

12. So we see that radioactivity keeps the earth's interior hot. After energy due to radioactivity eventually migrates to the earth's surface, in what form does it leave the earth?

TERRESTRIAL RADIATION

13. To get water from the ground, even in the hot desert, dig a hole about a half meter wide and a half meter deep. Place a cup at the bottom. Spread a sheet of plastic wrap over the hole and place stones along the edge to hold it secure. Weight the center of the plastic with a stone so it forms a cone shape. Why will water collect in the cup? (Physics can save your life if you're ever stranded in a desert!)



① Terr. Radiation makes water evapor.
② Evap. water trapped on plastic
③ Condensed water runs into cup.

(Some also will collect at night on the top of the sheet)

Print!

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ML Yell



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Materials Research Abstracts Library of Congress. Science and Technology Division, 1962

Low Dose Exposures in the Environment Christian Streffer, H. Bolt, D. Follesdal, P. Hall, J.G. Hengstler, P. Jacob, D Oughton, K. Prieß, E. Reh binder, E. Swaton, 2004-05-03 The ever increasing release of harmful agents due to human activities have led in some areas of the world to heavy pollution In order to protect human health and the environment environmental standards that shall limit the release and the concentration of those toxic agents in the environment and hence the exposure to it have to be established The related assessment and decision making procedures have to be based on solid scientific data about the effects and mechanisms of these agents as well as on ethical social and economic aspects For risk evaluation the knowledge of the dose response curve is an essential prerequisite Dose responses without a threshold dose are most critical in this connection Such dose responses are assumed for mutagenic and carcinogenic effects which therefore dominate also the discussion in this book In the environmentally important low dose range risk estimation can only be achieved by extrapolation from higher doses with measurable effects The extrapolation is accompanied with uncertainties which makes risk evaluation as well as risk communication frequently problematic In order to ensure rational efficient and fair decisions beyond a sound scientific assessment the dialogue between disciplines with the affected people and with the general public is necessary In this book the whole range of relevant and essential aspects of risk

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