

# Section 8 Review (Thompson)

10 pts

- Identify the correct full name of the acid described by the formula below.
- Calculate the molar mass of  $\text{H}_2\text{SO}_4$  (4.0 g/mol, 98.0 g/mol, 98.1 g/mol, 98.08 g/mol)



- Which of the following acids is a polyprotic acid?  $\text{HCl}$ ,  $\text{H}_2\text{O}$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}^{2+}$ ,  $\text{H}_2\text{O}$ , and  $\text{H}_2\text{O}$
- Which of the following acids is a polyprotic acid?  $\text{HCl}$ ,  $\text{H}_2\text{O}$ ,  $\text{H}_2\text{SO}_4$ ,  $\text{H}_2\text{O}^{2+}$ ,  $\text{H}_2\text{O}$ , and  $\text{H}_2\text{O}$
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1. 100%

2. 100%

3. 100%

4. 100%

5. 100%

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