PRACTICE PROBLEMS, CHAPTERS 1 - 3

(Covered from Ch. 3: Alkane and Alkyl Halide nomenclature only)

- 1. The atomic number of boron is 5. The correct electronic configuration of boron is:
 - A. 1s22s3
- B. 1s22p3
- C. 1s22s22p1
- D. 2s22p3
- E. 187287381
- 2. How many distinct p orbitals exist in the second electron shell, where n = 2?
- An oxygen atom has ______ valence electrons.
- 4. Draw a correct Lewis structure for boric acid, B(OH)3.
- Draw a correct Lewis structure for t-butanol, (CH3)3COH.
- Draw a correct Lewis structure for acetonitrile, CH3CN.
- 7. The formal charge on nitrogen in the compound below is _____

8. Which of the following are correct Lewis structures for nitric acid, HNO3?

- The formal charge on oxygen in dimethyl ether, CH3OCH3, is _______
- 10. Which of the following choices represent(s) a pair of resonance forms?

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R Sandford

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