

## \* Solid state \*

- \* **Solid** → have definite mass, volume, shape  
 → short intermolecular distance  
 → strong intermolecular force  
 → incompressible & rigid  
 → particle have fixed position & only oscillate about their position.

### \* solid have 2 types

#### crystalline

→ The solid which have definite order of arrangement of particle  
 ex. NaCl, Quartz

#### amorphous

→ The solid have irregular arrangement of particle  
 ex. glass, rubber, plastic, Quartz glass

properties	crystalline	Amorphous.
shape	definite geometric shape	irregular shape.
Melting	Melt at sharp temp.	Gradually soften over range of temp.
cleavage	when cut newly generated surface are plain & smooth	when cut new generated surface are irregular.
heat of fusion	definite heat of fusion	not have definite heat of fusion.
Anisotropy	Anisotropic (properties change in differ. direction)	isotropic (properties same in all direction)
Nature	True solid	Pseudo solid or super cooled liquid
order of arrangement	Long range order	only short range order.

# 12th Science Chemistry Target Solid State Notes

**LL Leslie**

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular gradient shape partially visible behind it.

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Treatise is to help define the field Perhaps the most general characteristic of solid state chemistry and one which helps differentiate it from solid state physics is its focus on the chemical composition and atomic configuration of real solids and on the relationship of composition and structure to the chemical and physical properties of the solid Real solids are usually extremely complex and exhibit almost infinite variety in their compositional and structural features

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