

Limiting Reactant and Percent Yield Practice

Name _____

- 1) Consider the following reaction:



Which reactant is limiting, assuming we started with 30.0 grams of ammonium nitrate and 50.0 grams of sodium phosphate. What is the mass of each product that can be formed? What mass of the excess reactant(s) is left over?

- 2) Consider the following reaction:



Which reactant is limiting, assuming we start with 100. grams of calcium carbonate and 45.0 grams of iron (III) phosphate. What is the mass of each product that can be formed? What mass of the excess reactant(s) is left over?

12 3 Limiting Reagent Extra Practice

Terry C. Jones



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