

01. CHEMICAL REACTIONS and EQUATIONS

WORKSHEET 1

- (1.) Which statement is correct about the following reaction?



- (1) ZnO is being oxidized
 - (2) CO is being reduced
 - (3) CO_2 is being oxidized
 - (4) ZnO is being reduced
- (2.) Consider the following reaction :
 $x\text{C}_2\text{H}_6(\text{g}) + y\text{O}_2(\text{g}) \rightarrow m\text{CO}_2(\text{g}) + n\text{H}_2\text{O}(\text{l})$
Which of the following set of coefficients balanced the above redox reaction? x y m n
- (1) 1 3 2 2
 - (2) 2 7 4 6
 - (3) 2 3 2 2
 - (4) 1 7 2 3
- (3.) Identify the type of chemical reaction.
 $\text{H}_2\text{O}_2(\text{l}) \xrightarrow{\text{I}^-} \text{H}_2\text{O}(\text{l}) + \text{O}_2(\text{g})$
- (1) Neutralization reaction
 - (2) Combination reaction
 - (3) Decomposition reaction
 - (4) Double Displacement reaction
- (4.) Which of the given equations exhibit a combination reaction involving two compounds?
- (1) $2\text{CO} + \text{O}_2 \rightarrow 2\text{CO}_2$
 - (2) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$
 - (3) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$
 - (4) $\text{CaO} + \text{H}_2\text{O} \rightarrow \text{Ca(OH)}_2$
- (5.) In the equation,
 $\text{Na}_2\text{CO}_3 + x\text{HCl} \rightarrow 2\text{NaCl} + \text{CO}_2 + \text{H}_2\text{O}$, the value of x is:
- (1) 1
 - (2) 2
 - (3) 3
 - (4) 4

- (6.) What will be the value of x, y and z in following equation?



- (1) $x = 1, y = 2, z = 2$
 - (2) $x = 2, y = 2, z = 2$
 - (3) $x = 2, y = 2, z = 1$
 - (4) $x = 1, y = 2, z = 4$
- (7.) The main cause of rancidity in foods is:
- (1) bacteria
 - (2) proteins
 - (3) antioxidants
 - (4) oxidation of the fatty acid molecule
- (8.) Constituents of the compounds can be separated by
- (1) physical methods
 - (2) chemical reactions
 - (3) Both A and B
 - (4) None of A or B
- (9.) Which of the following oxide(s) of iron would be obtained on prolonged reaction of iron with steam?
- (1) FeO
 - (2) Fe_3O_4
 - (3) Fe_2O_3
 - (4) Fe_2O_3 and Fe_3O_4
- (10.) Which black substance is formed when silver is kept open for few days?
- (1) Silver sulphate
 - (2) Silver sulphide
 - (3) Silver sulphite
 - (4) Silver oxide
- (11.) Oxidation number of S in S^{2-} is
- (1) -2
 - (2) 0
 - (3) -6
 - (4) +6
- (12.) Which of the following is a decomposition reaction?

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