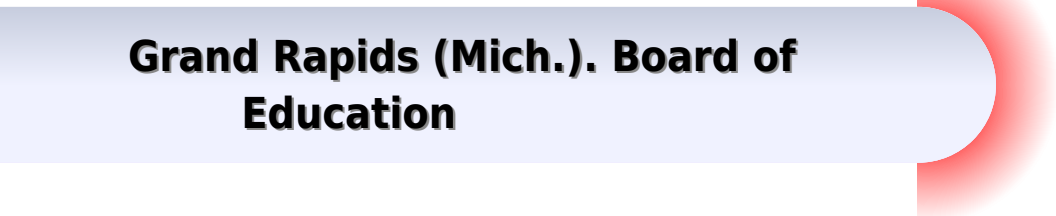


1. What is the correct electron dot diagram for the aluminum atom in the ground state?
(1) $\text{Al} \cdot$ (2) $\text{Al} \cdot$ (3) $\cdot \text{Al} \cdot$ (4) $\cdot \text{Al} \cdot$
2. Which pair of atoms are isotopes of element X?
(1) ${}_{90}\text{X}^{226}$ and ${}_{91}\text{X}^{226}$ (2) ${}_{91}\text{X}^{227}$ and ${}_{90}\text{X}^{227}$ (3) ${}_{90}\text{X}^{226}$ and ${}_{91}\text{X}^{227}$ (4) ${}_{90}\text{X}^{227}$ and ${}_{90}\text{X}^{226}$
3. A binary compound of sodium is
(1) sodium chlorate (2) sodium chlorite (3) sodium perchlorate (4) sodium chloride
4. When the vapor pressure in an open container equals the atmospheric pressure, the liquid will
(1) boil (2) freeze (3) crystallize (4) melt
5. When the equation, $___\text{Na(s)} + ___\text{H}_2\text{O(l)} \rightarrow ___\text{NaOH(l)} + ___\text{H}_2\text{(g)}$ is correctly balanced using smallest whole numbers, the coefficient of water will be
(1) 1 (2) 2 (3) 3 (4) 4
6. An ionic bond forms between atoms of
(1) I and Cl (2) K and Cl (3) P and Cl (4) H and Cl
7. The greatest absorption of energy occurs as an electron moves from
(1) 1s to 3s (2) 3p to 3s (3) 4d to 4s (4) 4s to 3p
8. What is the total number of valance electrons in an atom with a total of 13 electrons?
(1) 1 (2) 2 (3) 3 (4) 4
9. Which of the following has the least mass?
(1) proton (2) neutron (3) hydrogen nucleus (4) electron
10. What is the total number of sublevels in the second principal energy level?
(1) 1 (2) 2 (3) 3 (4) 4
11. Which of the following atoms has the greatest tendency to gain electrons?
(1) Al (2) Rb (3) I (4) F
12. Which is the formula for ammonium acetate?
(1) $\text{NH}_4\text{CH}_3\text{COO}$ (2) NH_3CO_3 (3) NaCH_3COO (4) $\text{CH}_3\text{COONH}_4$
13. What type of bond is present in copper wire?
(1) covalent (2) ionic (3) electrovalent (4) metallic
14. Which type of bond is present in a water molecule?
(1) polar covalent (2) nonpolar covalent (3) ionic (4) metallic
15. As sodium reacts with fluorine to form the compound NaF, each sodium atom will
(1) gain 1 electron (2) gain 2 electrons (3) lose 1 electron (4) lose 2 electrons
16. What is the proper name of the compound with the formula, H_2SO_4 ?
(1) hydrosulfuric acid (2) hydrosulfurous acid (3) sulfuric acid (4) sulfurous acid

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Table of Contents **10th Grade Chemistry Review**

1. Understanding the eBook 10th Grade Chemistry Review
 - The Rise of Digital Reading 10th Grade Chemistry Review
 - Advantages of eBooks Over Traditional Books
2. Identifying 10th Grade Chemistry Review
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 10th Grade Chemistry Review
 - User-Friendly Interface
4. Exploring eBook Recommendations from 10th Grade Chemistry Review
 - Personalized Recommendations
 - 10th Grade Chemistry Review User Reviews and Ratings
 - 10th Grade Chemistry Review and Bestseller Lists

5. Accessing 10th Grade Chemistry Review Free and Paid eBooks
 - 10th Grade Chemistry Review Public Domain eBooks
 - 10th Grade Chemistry Review eBook Subscription Services
 - 10th Grade Chemistry Review Budget-Friendly Options
6. Navigating 10th Grade Chemistry Review eBook Formats
 - ePub, PDF, MOBI, and More
 - 10th Grade Chemistry Review Compatibility with Devices
 - 10th Grade Chemistry Review Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 10th Grade Chemistry Review
 - Highlighting and Note-Taking 10th Grade Chemistry Review
 - Interactive Elements 10th Grade Chemistry Review
8. Staying Engaged with 10th Grade Chemistry Review
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 10th Grade Chemistry Review
9. Balancing eBooks and Physical Books 10th Grade Chemistry Review
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 10th Grade Chemistry Review
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 10th Grade Chemistry Review
 - Setting Reading Goals 10th Grade Chemistry Review
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 10th Grade Chemistry Review
 - Fact-Checking eBook Content of 10th Grade Chemistry Review
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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