

Answer Key

Chapter 10

Practice Problems

1. 9.45×10^{24} molecules CO_2
2. 3.27×10^{22} molecules H_2O
3. 1.53 mol Fe
4. 0.0341 mol sucrose
5. 1.08×10^{21} formula units CuSO_4
6. 839 g Sb
7. 5.53 g Se
8. 6.95 mol S
9. 12.5 mol He
10. 0.0213 mol Ni
11. 6.17×10^{21} atoms Pt
12. 1.88×10^{28} atoms S
13. 39.7 g Hg
14. 6.35×10^{-3} g
15. 4.36×10^{16} Pb ions
16. 0.220 mol C_6H_6
17. 2.856 mol KClO_3
18. 40.2 g SnSO_4
19. 3.40×10^2 g (340 g expressed to 3 significant figures)
20. 13.1 g Zn and 37.5 g $\text{Cu}(\text{NO}_3)_2$
21. 52.93% Al, 47.07% O
22. 16.39% Mg, 18.89% N, 64.72% O
23. 39.17% O
24. 49.30% N
25. The percent compositions will be the same because the ratio of carbon atoms to hydrogen atoms is the same, 1:1.
26. CH_2O
27. ZnN_2O_6
28. Al_4C_3
29. $\text{C}_3\text{H}_4\text{O}_3$
30. $\text{C}_8\text{H}_8\text{N}_2\text{O}_2$
31. $\text{B}_3\text{H}_6\text{N}_3$
32. $\text{Ag}_2\text{C}_2\text{O}_4$
33. P_2S_5 ; same as empirical formula
34. $\text{C}_9\text{H}_{12}\text{N}_6$
35. $\text{CeI}_3 \cdot 9\text{H}_2\text{O}$
36. $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$
37. $\text{Pb}(\text{C}_2\text{H}_3\text{O}_2)_2 \cdot 3\text{H}_2\text{O}$
38. $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$

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Anu Sharma



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Calculating Theoretical Yield Is Not Impossible! Michael S. Leonard, 2025-02-25 Calculating theoretical and percent yield is a fundamental skill for the laboratory. This book primarily targets Organic Chemistry Laboratory courses at the high school or college and university level as a supplemental resource to help students master this skill. It begins with simple examples from everyday life, demonstrates the importance of balancing the equation, addresses the role of the mole in these computations, discusses different types of liquids, considers the role of significant figures, and culminates with the planning of syntheses. There are suggestions for further reading as well as practice problems and questions to ensure mastery. Begins with examples from everyday life that enable students to understand the concepts of theoretical and percent yield before applying those concepts to the laboratory. Addresses the necessity of balancing the reaction equation, the centrality of the mole in these calculations, and the role of significant figures in reporting the answer. Explains how to approach the calculations when using neat liquids or solutions. The culmination of this text is the use of the same thought processes to plan the amounts of reactants needed for syntheses of desired quantities of product. All of the problems in the book include detailed solutions with accompanying text to explain the answers, and ancillaries also include suggestions for further reading.

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concepts developed in the chapters This book is designed for introductory courses in analytical chemistry especially those shorter courses servicing chemistry majors and life and health science majors *Solid-State Chemistry* Ashok Kumar Jha, 2023-08-04 Solid state reactions have been growing in importance in the preparation of solids crystal growth and single crystals of elements In this new volume the author discusses in depth recent advances in solid state chemistry the latest techniques of characterization of solids and several new dimensions of solid state problems Solid state reactions their kinetics phase transitions and magnetic properties are discussed in detail along with electrical properties semiconductors metals and insulators with examples and explanations Dielectric piezoelectric and ferroelectric properties of solids have been considered as well Different theories of catalysis and adsorption isotherms are covered and the author includes several experimental findings as well And of course preparation of new solids by taking advantage of recent experimental techniques have also been included in this volume This volume provides a complete description of solid state reactions electrical conductivity characterization of solids organic solids and intercalation between layers of solids The band theories of metals semiconductors and insulators are covered In characterization of solids TGA DTA DSC FTIR and PXRD techniques have been described Preparation and properties of organic solids are also explained Dielectric properties photovoltaic effects piezoelectricity ferroelectricity and antiferroelectricity have been dealt with giving relatable examples A number of excitation energy levels giving rise to luminescence and photoluminescence have been explained under the topic optical processes Diamagnetic paramagnetic and ferromagnetic behaviors of solids are also explained **Target IIT JEE (A complete solution in Mathematics) Class XI** Umesh Jalan, 2021-04-16 The key notes and questions present in this book have been tested by millions of IIT JEE students over the years This book contains all the important and frequent ask concept which is drive from several notes an previous year paper of JEE AIPMT JIPMER AIIMS NEET and various state engineering and medical entrance examinations Even a below average student can crack JEE after doing this book *Chemistry* John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant Chapter-wise DPP Sheets for Chemistry JEE Main Disha Experts, The book Chapter wise Daily Practice Problem DPP Sheets for Chemistry JEE Main contains 1 Carefully selected Questions 30 per DPP in Chapter wise DPP Sheets for Practice At the end one Full Test is provided 2 The book is divided into 30 Chapter wise DPPs based on the NCERT 3 Time Limit Maximum Marks Cutoff Qualifying Score for each DPP Sheet is provided 4 These sheets will act as an Ultimate tool for Concept Checking Speed

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