PAPER 2 2 hours

ESSAY

[100 MARKS]

Answer four questions in all: question 1 in Section A and three questions from Section B.

All questions carry equal marks.

No marks will be awarded for answering questions **not** specified for your own country. Credit will be given for clarity of expression and orderly presentation of materials.

SECTION A

Answer all the questions in this section.

				The state of the s
1.	(a)	(1)	Define an acid according to the Lewis concept.	
		(iii)	Give one example of a Lewis acid.	[3 marks]
	(b)	Explain salting out in soap preparation.		[2 marks]
	(C)	State the reagent and condition necessary for the following conversion:		
			$C \equiv C - H - \blacktriangleright Ag - C \equiv C - Ag$	[2 marks]
	(cl)	What is the percentage abundance of an isotope?		[2 marks]
	(c)	(E)	Why does the element with atomic number 18 not have on oxide?	
		(iii)	Explain why chlorine (I) oxide has a low melting point.	[3 marks]
	(f)		cribe a test to distinguish between concentrated of and concentrated H ₂ SO ₄ ,	[3 marks]
	(g)	State two differences between an electrochemical cell and an electrolytic cell.		[3 marks]
	(h)		does the trend in ionization energy affect the tivity of group 1 elements?	[3 marks]
	(i)	Defin	ne the term molecular formula.	[2 marks]
	(i)		ate which of the gases 11, and NH, would deviate from ideal behaviour.	
		(iii) G	ive reasons for the answer stated in 1() (i)	[3 marks]

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