

Gas Variables

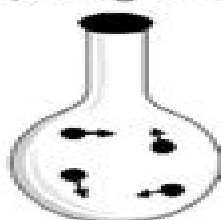
How are the variables that describe a gas related?

Why?

Imagine buying a balloon bouquet at a party store. How will the helium gas in the bouquet behave if you carry it outside on a hot summer day? How will it behave if you carry it outside during a snowstorm? What happens if the balloons are made of latex, which can stretch? What happens if the balloons are made of Mylar[®], which cannot stretch? What if you add just a small amount of gas to each balloon? What if you add a lot of gas? In this activity, you will explore four variables that quantify gases—pressure (P), volume (V), temperature (T), and moles (n) of gas. These four variables can be related mathematically so that predictions about gas behavior can be made.

Model 1 – Gases in a Nonflexible Container

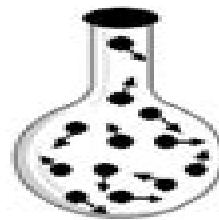
Experiment A (Adding more gas)



A1
Volume = 1 unit
External pressure = 1 atm
Internal pressure = 1 atm
Temperature = 200 K

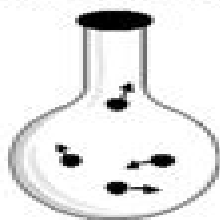


A2
Volume = 1 unit
External pressure = 1 atm
Internal pressure = 2 atm
Temperature = 200 K

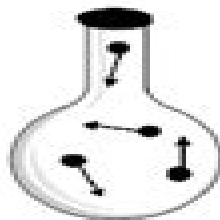


A3
Volume = 1 unit
External pressure = 1 atm
Internal pressure = 3 atm
Temperature = 200 K

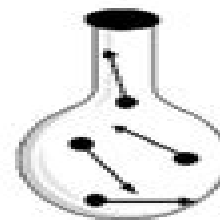
Experiment B (Heating the gas)



B1
Volume = 1 unit
External pressure = 1 atm
Internal pressure = 1 atm
Temperature = 200 K



B2
Volume = 1 unit
External pressure = 1 atm
Internal pressure = 2 atm
Temperature = 400 K



B3
Volume = 1 unit
External pressure = 1 atm
Internal pressure = 3 atm
Temperature = 600 K

*Note: Volume in this model is recorded in units rather than liters because 4 molecules of gas at the conditions given would occupy a very small space ($\sim 1 \times 10^{-20}$ L). The particles shown here are much larger compared to the space between them than actual gas particles.

Gas Variables Packet Answers

R Pring



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Gas World ,1971 Oil and Gas Journal Petroleum Publishing Company,1984 **Causality, Measurement Theory and the Differentiable Structure of Space-Time** R. N. Sen,2010-02-11 Introducing graduate students and researchers to mathematical physics this book discusses two recent developments the demonstration that causality can be defined on discrete space times and Sewell s measurement theory in which the wave packet is reduced without recourse to the observer s conscious ego nonlinearities or interaction with the rest of the universe The definition of causality on a discrete space time assumes that space time is made up of geometrical points Using Sewell s measurement theory the author concludes that the notion of geometrical points is as meaningful in quantum mechanics as it is in classical mechanics and that it is impossible to tell whether the differential calculus is a discovery or an invention Providing a mathematical discourse on the relation between theoretical and experimental physics the book gives detailed accounts of the mathematically difficult measurement theories of von Neumann and Sewell *Food Processing Industry* ,1969 Scientific and Technical Aerospace Reports ,1980 *English Mechanic and Mirror of Science* ,1906 *English Mechanic and World of Science* ,1906 **Resources in Education** ,1997 **English Mechanic and Mirror of Science and Art** ,1906 **Illinois Chemistry Teacher** ,1998 **Research in Education** ,1973 **Resources in Education** ,1996 Product Testing Activities ,1993 Contains activities in which students make practical use of their knowledge of science and technology to test the quality of a variety of consumer goods Encourages students to make intelligent choices as consumers **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science** ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **The Journal of Canadian Petroleum Technology** ,1984 **Commerce Business Daily** ,1999 **World Oil** ,1984 **Petroleum Engineer International** ,1984 **Journal of Petroleum Technology** ,1983 El-Hi Textbooks in Print ,1976

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