

Electricity and Magnetism Study Guide

Due 12/17/2019

- 1.) What causes a balloon to be drawn towards/stick to a sweater? What causes your hair to stand as you bring a charged balloon towards it?

-The balloon and the sweater/hair have opposite charges

- 2.) Describe the charges of the pictures below:



The objects are attracted to each other (opposite charge)



The objects are repelling each other (same charge)

- 3.) Fill in the blanks for the following sentence: Materials with opposite charges are attracted to each other while materials with the same charge are repelled from each other.

- 4.) After walking across the carpet to open your bedroom door you feel a shock on your hand. What caused the shock?

-Discharge of electricity from your hand

- 5.) What is friction? Provide an example

-At item becoming charged by rubbing against another item. Ex: Clothes sticking together in dryer, rubbing your feet against the carpet/rug

- 6.) What is conduction? Provide an example

-An item becoming charged through direct contact. Ex:

- 7.) What is induction? Provide an example

-An item charging through indirect contact. Ex: Rubbing a balloon against your leg then holding it near your hair and your hair sticks up

- 8.) How does an electroscope work?

- 9.) What is a conductor?

-Allows electrons to move more easily

- 10.) What is an insulator?

-Does NOT allow electrons to move easily

- 11.) An object is magnetized when its domains are aligned.

- 12.) If the poles of two magnets are attracted to each other what can you infer?

-They are opposites (one is north pole one is a south pole)

- 13.) What is a solenoid?

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web aug 30 2022 acid base titrations are usually used to find the amount of a known acidic or basic substance through acid base reactions the analyte titrand is the solution with an unknown molarity the reagent titrant is the solution with a known molarity that will react with the analyte

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web conclusion the purpose of the experiment was to identify the unknown solid acid 17 by calculating its molar mass this was done by titration of the acid with sodium hydroxide in order to find the volume of sodium hydroxide required to reach the end point and that value was used in unit analysis to find the molar mass and identify the unknown solid acid

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web by utilizing the half equivalence point on the titration curve the pka of the acid analyte was determined to be 3.655 furthermore using the stoichiometric ratio between the naoh and the acid analyte 1:1 we could experimentally determine the number of moles of the acid in the solution to be 0.00731 moles making the experimental molar mass

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 $\text{HCl(aq)} + \text{NaOH(aq)} \rightarrow \text{NaCl(aq)} + \text{H}_2\text{O(l)}$
one mole of HCl would be fully neutralized by one mole of NaOH

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standardization of sodium hydroxide NaOH and oxalic acid $\text{H}_2\text{C}_2\text{O}_4 \cdot 2\text{H}_2\text{O}$ and the molar mass of an unknown solid acid using the molar mass of the standardized sodium hydroxide solution

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moles of naoh na added is equal to the number of moles of khp nb in the

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