

Nuclear Chemistry

Section 25.1 Nuclear Radiation

In your textbook, read about the terms used to describe nuclear changes.

Use each of the terms below just once to complete the passage.

alpha particle
beta particles

radioactivity
radiation

gamma ray
X ray

radioisotope
radioactive decay

The discovery of the (1) _____ in 1895 by Wilhelm Roentgen opened a whole new field of research. Among those who worked in this new field were Pierre and Marie Curie. The Curies discovered that some forms of matter give off

(2) _____, a combination of particles and energy. Marie Curie named this process (3) _____. Another term used to describe the process by which one element spontaneously changes into another element is (4) _____. Any isotope that undergoes such changes is called a(n) (5) _____.

There are three common forms of radiation. One type is a form of energy known as (6) _____. The other types of radiation consist of particles. The form of radiation containing the heavier particle is made up of helium nuclei called (7) _____. The form of radiation containing the lighter particle consists of electrons called (8) _____.

In your textbook, read about the discovery of radioactivity.

Complete each statement.

9. Wilhelm Roentgen discovered the form of energy known as _____.
10. The form of nuclear radiation that has the greatest penetrating power is the _____.
11. When a radioactive nucleus gives off a gamma ray, its atomic number increases by _____.
12. The three types of radiation were first identified by _____.
13. Each alpha particle carries an electric charge of _____.
14. Each beta particle carries an electric charge of _____.
15. Each gamma ray carries an electric charge of _____.

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