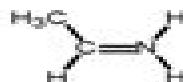


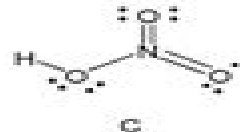
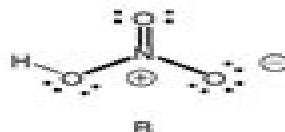
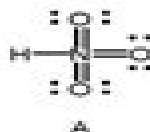
## PRACTICE PROBLEMS, CHAPTERS 1 - 3

(Covered from Ch. 3: Alkane and Alkyl Halide nomenclature only)

- The atomic number of boron is 5. The correct electronic configuration of boron is:  
 A.  $1s^2 2s^3$       B.  $1s^2 2p^3$       C.  $1s^2 2s^2 2p^1$       D.  $2s^2 2p^3$       E.  $1s^2 2s^2 3s^1$
- How many distinct  $p$  orbitals exist in the second electron shell, where  $n = 2$ ?
- An oxygen atom has \_\_\_\_\_ valence electrons.
- Draw a correct Lewis structure for boric acid,  $B(OH)_3$ .
- Draw a correct Lewis structure for  $t$ -butanol,  $(CH_3)_3COH$ .
- Draw a correct Lewis structure for acetonitrile,  $CH_3CN$ .
- The formal charge on nitrogen in the compound below is \_\_\_\_\_.



- Which of the following are correct Lewis structures for nitric acid,  $HNO_3$ ?



- The formal charge on oxygen in dimethyl ether,  $CH_3OCH_3$ , is \_\_\_\_\_.
- Which of the following choices represent(s) a pair of resonance forms?



A



B



C

# 4 Practice Problems Chemistry

**Julia Burdge**



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