

MODEL QUESTION PAPER - 2

CHEMISTRY

(SCIENCE PAPER - 2)

Maximum Marks : 60

Time allowed : Two hours

Answers to this Paper must be written on the paper provided separately.

You will not be allowed to write during first 15 minutes.

The time given at the head of this Paper is the time allowed for writing the answers.

Section A is compulsory. Attempt any four questions from Section B.

The intended marks for questions or parts of questions are given in brackets [].

SECTION A (40 Marks)

Attempt all questions from this Section.

Question 1

(a) Choose the correct answers from the given options :

[15]

(i) $2H$ stands for :

- A. One atom of hydrogen gas
- B. One molecule of hydrogen gas
- C. Two molecules of hydrogen gas
- D. Two atoms of hydrogen gas

(ii) During an endothermic reaction :

- A. The temperature of surroundings increases.
- B. The temperature of surroundings decreases
- C. The heat in the surroundings increases
- D. The reactants melt.

(iii) If pressure is doubled for a fixed mass of a gas, its volume will become :

- A. 4 times
- B. $1/2$ times
- C. 2 times
- D. No change

(iv) When hydrochloric acid is added to silver chloride solution :

- A. No reaction takes place.
- B. A white precipitate is formed.
- C. A white solution is obtained.
- D. A reddish brown precipitate is formed.

(v) Elements X, Y and Z have atomic numbers 6, 9 and 12 respectively. Which of them will form a cation ?

- A. Z
- B. Y
- C. X
- D. Depends on the reactivity of the elements.

(vi) The atomic numbers of elements A, B, C and D are 2, 6, 12 and 20 respectively. Which of them has 2 electrons in its N shell ?

- A. A
- B. D
- C. B
- D. C

(vii) The number of atoms in a chloride ion is :

- A. One
- B. Two
- C. Three
- D. Four

(viii) The acidic radical in Sulphurous acid is :

- A. H^+
- B. SO_3^{2-}
- C. SO_4^{2-}
- D. SO_3^{2-}

(ix) When an oxygen ion is converted to an oxygen atom, it undergoes :

- A. Reduction
- B. Redox reaction
- C. Oxidation
- D. Combination reaction

(x) The element whose atomic weight was corrected on the basis of its position in Mendeleev's periodic table is :

- A. Hydrogen
- B. Gold
- C. Silver
- D. Copper

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