

CHEMISTRY HSSC-I
SECTION – A (Marks 17)

Time allowed: 25 Minutes

Note: Section-A is compulsory. All parts of this section are to be answered on the separately provided OMR Answer Sheet which should be completed in the first 25 minutes and handed over to the invigilator. Do not use lead pencil.

Q-1: Encircle the correct option i.e. A/B/C/D. Each part carries 1 mark.

- (i) What is the mass of 4.48×10^{21} atoms of magnesium?
- | | |
|---------------|-------------|
| A. 0.20 g Mg | B. 1.2 g Mg |
| C. 0.178 g Mg | D. 20 g Mg |
- (ii) The values of magnetic quantum number give us information about the:
- | | |
|------------------------------------|---------------------------|
| A. Probability of finding electron | B. Orientation of orbital |
| C. No of shells | D. All of these |
- (iii) A sub-shell with $n = 6$, $l = 2$ can accommodate a maximum of:
- | | |
|-----------------|-----------------|
| A. 12 electrons | B. 36 electrons |
| C. 10 electrons | D. 72 electrons |
- (iv) Geometry of molecule will be pyramidal if the outer most shell of the central atom has?
- | | |
|------------------------------|----------------------------|
| A. 3 bond pair one lone pair | B. 2 bond pair 2 lone pair |
| C. 1 bond pair 3 lone pair | D. 3 lone pair 1 bond pair |
- (v) Under what conditions will a gas most closely follow ideal behavior?
- | |
|---------------------------------------|
| A. Low temperature and low pressure |
| B. High pressure and high temperature |
| C. High pressure and low temperature |
| D. Low pressure and high temperature |
- (vi) 80.0 liters of oxygen is collected over water at 50.0 °C. The atmospheric pressure in the room is 96.00 kPa. What is the partial pressure of the oxygen?
- | | |
|--------------|--------------|
| A. 29.23 kPa | B. 83.66 kPa |
| C. 93.05 kPa | D. 79.70 kPa |

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