

# Chemistry Practice Worksheet Chemical Formulas

Write a chemical formula for each compound.

|     |                                                |                   |
|-----|------------------------------------------------|-------------------|
| 1.  | <u><math>\text{Al}_2\text{O}_3</math></u>      | aluminum oxide    |
| 2.  | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 3.  | <u><math>\text{H}_2\text{S}</math></u>         | hydrogen sulfide  |
| 4.  | <u><math>\text{K}_2\text{SO}_4</math></u>      | potassium sulfate |
| 5.  | <u><math>\text{H}_2\text{CO}_3</math></u>      | carbonic acid     |
| 6.  | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 7.  | <u><math>\text{Ca}_3(\text{PO}_4)_2</math></u> | calcium phosphate |
| 8.  | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 9.  | <u><math>\text{Ca}_3(\text{PO}_4)_2</math></u> | calcium phosphate |
| 10. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 11. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 12. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 13. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 14. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 15. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
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| 19. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 20. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
| 21. | <u><math>\text{H}_2\text{SO}_4</math></u>      | sulfuric acid     |
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| 79. | <u><math>\text{H}_2\text{SO}_4</math></u> | sulfuric acid |
| 80. | <u><math>\text{H}_2\text{SO}_4</math></u> | sulfuric acid |

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## Key

**E Durkheim**



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