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# Chemistry Practice Worksheet: Chemical Formulas

Write a chemical formula for each compound:

1.  $Al_2O_3$  aluminum oxide
2.  $Hg_2Cl_2$  mercurous chloride
3.  $H_2SO_4$  sulfuric acid
4.  $K_2SO_4$  potassium sulfate
5.  $CH_3COOH$  acetic acid
6.  $AgNO_3$  silver nitrate
7.  $Ca(OH)_2$  calcium hydroxide
8.  $Na_2CO_3$  sodium carbonate
9.  $CaCl_2$  calcium chloride
10.  $NaOH$  sodium hydroxide
11.  $H_2SO_3$  sulfurous acid
12.  $NaNO_3$  sodium nitrate
13.  $FeCl_3$  ferric chloride
14.  $PbCl_2$  lead(II) chloride
15.  $FeCl_2$  ferrous chloride
16.  $Ca_3(PO_4)_2$  calcium phosphate
17.  $Na_2S$  sodium sulfide
18.  $Ca_3(PO_4)_2$  calcium phosphate
19.  $CH_4$  methane
20.  $NaCl$  sodium chloride
21.  $Fe_2(SO_4)_3$  ferric sulfate
22.  $H_2SO_4$  sulfuric acid
23.  $CaSO_4$  calcium sulfate
24.  $Na_2CO_3$  sodium carbonate
25.  $Al_2(SO_4)_3$  aluminum sulfate
26.  $Ag_2O$  silver oxide
27.  $CO$  carbon monoxide
28.  $K_2CO_3$  potassium carbonate
29.  $HNO_3$  nitric acid
30.  $CaO$  calcium oxide
31.  $Na_2C_2O_4$  sodium oxalate
32.  $Na_2C_2O_4$  sodium oxalate
33.  $Na_2C_2O_4$  sodium oxalate
34.  $Na_2C_2O_4$  sodium oxalate
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**Wilder Dwight Bancroft**



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