- Alkyne: Alkynes are unsaturated hydrocarbons in which two carbon atoms are connected by a triple bond. The general formula of alkynes is C_nH_{2n-2}. Alkynes are also called acetylenes. Examples: Propyne, Pentyne, etc.
- Homologous series: Homologous series is a group of organic compounds having similar structures and similar chemical properties in which the successive compounds differ by CH₂ group. All members of a homologous series share a general formula.
- Methane: It is the smallest aliphatic hydrocarbon. Its molecular formula is CH₄. Methane gas is found in petroleum mines, swamps and in biogas. It is used as a source of heat in the form of LPG.
- Ethane: It is a saturated hydrocarbon having two carbon atoms. It is the second member of alkane series. It is found in petroleum mines and used as a source of heat in the form of natural gas or gobar gas.
- Propane: It is a saturated hydrocarbon having three carbon atoms. It is the third member of alkane series. Its molecular formula is C₂H₈. It is a colourless gas found in petroleum mines. It is used as fuel in lighters.
- Butane: It is a saturated hydrocarbon having four carbon atoms. This gas is found in petroleum mine and used in LP gas along with methane.
- Isomers and Isomerism: Isomers are the organic compounds having the same molecular formula but different structures and properties. Isomerism is defined as the existence of two or more different organic compounds having the same molecular formula but different structures and properties.
- Alcohol: Organic compound containing hydroxyl group (-OH) is called alcohol. Example: Methyl Alcohol [CH₂OH].
- Glycerol: Glycerol is a colourless thick liquid having sweet taste. It is a trihydric alcohol, its molecular formula is C₃H₅(OH)₃.

In order to meet the outcomes of this unit, study the following model questionsanswers and solutions to numerical problems deeply for practice.

Subject Matter

Section 'A'

(1 Mark Each)

Multiple Choice Questions

- Identify organic compounds from the given compounds. (K)
 - a. CaCO₃

b. H₃CCH₃

c. CO2

- d. C₂H₅OH
- Which of the given hydrocarbons belongs to alkane group? (HA)
 - a. Propyne

b. Ethene .

g/Butane

- d. Butene
- Which of the following is the correct structural formula of butene? (U)

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