Chapter 22 Assignment Answers - Organic Chemistry Unit

Chapter 22: 37 - 41, 50, 53, 55, 56, 59, 66, 73

37. pentane: CH₃CH₂CH₂CH₂CH₃

hexane: CH₃CH₂CH₂CH₂CH₂CH₃

38. (a) propane (b) octane (c) pentane

 Remember... "alkyl" groups are side groups that branch off a longer-chain hydrocarbon.

methyl - CH₃ ethyl - CH₂CH₃ propyl - CH₂CH₂CH₃

- 40. (a) 2-methylbutane
 - (b) 2,3-dimethylbutane
 - (c) 3-methylhexane
- 41. Alkane molecules are nonpolar because all of the bonds within the molecule are nonpolar bonds. You can make an alkane molecule polar by adding a functional group to it, however.

50. Both structures are the same, because the alternating double bonds don't really exist! Remember that the carbon-to-carbon bonds in benzene are all identical. CH_3 CH_3

 CH_3

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Laurie Sorge

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