General Instructions:

Read the following instructions very carefully and strictly follow them:

- (i) This question paper comprises 15 questions. All questions are compulsory.
- (ii) This question paper is divided into three sections A, B and C.
- Section A Questions No. 1 to 7 are short answer type questions. Each question carries 2 marks.
- Section B Questions No. 8 to 13 are also short answer type questions. Each question carries 3 marks.
- Section C Questions No. 14 and 15 are case-based questions. Each question carries 4 marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

SECTION A

- 1. (a) Write the molecular formula of the following carbon compounds:
 - (i) Methane
 - (ii) Propane
 - (b) Carbon compounds have low melting and boiling points. Why?
- The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.
 - (a) Write the group numbers of these elements in the Modern Periodic Table.
 - (b) Write the molecular formula of the compound formed when X and Y combine with each other.
 - (c) Which of the two is electropositive?
- 3. (a) Which of the following flowers will have higher possibility of self-pollination?

Mustard, Papaya, Watermelon, Hibiscus

- (b) List the two reproductive parts of a bisexual flower.
- 4. Which one of the two multicellular organisms Spirogyra and Planaria reproduces by regeneration and why? Give an example of any other organism which can also reproduce by the same process.

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