First PUC Supplimentary Examination, April - 2018

Subject : CHEMISTRY

Commence of the last of the la	3 Hours 15 Min	Max. Marks: 7
i)		wer all the parts.
-1/		
	c) Part-C carries 15 marks.	
No.	d) Part-D carries 35 marks.	
iii)	Write neat labelled diagrams wherever required.	
iv)	Use only simple calculators. (Use of Scientific Calculator is not allowed.)	
(V)	Write balanced chemical equations wherever required.	THE REAL PROPERTY.
vi)	Use log tables if necessary.	A COLUMN TO A COLU
	Part - A	
LAnsv	ver the following questions:	$10 \times 1 = 1$
1.	Class the Coloration secretary of 222 600	
-2.	What is the value of compressibility factor (z) for an ideal gas?	
3.	What is common ion effect?	
4.	Which element has highest electronegativity?	
5.	Find the oxidation number of 'P' in NaH,PO.	
6.		
7.	What is Producer Gas?	
8.	Give an example for Zeolite.	
9.	Write the functional group of Carboxylic acids.	
10.	Identify the product 'A'.	
	OH	
	☐ Zn heat A + ZnO	
	O	
	Part - B	
II.Ans	wer any Five of the following questions. Each question carries two marks :	$5 \times 2 = 1$
11.	Give example for :	adillera a service a
	a) Element b) Compound	
12.	Write any two differences between Ideal gas and Real gas.	
13.	What is Dipole Moment? Give its unit.	
14.	Mention any two similarities between Lithium and Magnesium (Mg).	
	Mention any two similarities between Lithium and Magnesium (Mg).	
15.	What is the action of cone. H ₂ SO ₄ on formic acid at 373K? Give equation.	
16.	Explain Markovnikov's rule with an example.	
17.	Stat and explain Huckel's rule of Aromaticity.	
18.	W BILL IS ENCILLY CHIVE HS STUBILICABOR.	
	All them (CHI has 14" O'AT February CHI O'AN DAYO Respirators to accommission of amounts)	
	positioners and not be said Part - C to be	
III.An	swer any Five of the following questions:	$5 \times 3 = 1$
19.	What is Electron Gain Enthalpy? How does it vary along the period and down	
20.	Explain SP hybridisation in Ethyne (C,H,).	
21.	Write the electronic configuration of Oxygen molecule using Molecular orbital	theory and evalui
	its paramagnetic property.	theory and explain
22.	Define the terms :	
-		
	a) Bond Enthalpy	
	b) Bond Angle	FRESTWY OF
	c) Bond Length	THE THE PARTY AND THE PARTY AN

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