

2.5: Practice - Alkane Nomenclature

I. The parent compound must have the longest chain of carbon atoms

Learn: methane (1 carbon atom), ethane (2 carbon atoms), propane (3 carbon atoms), butane (4 carbon atoms), pentane (5 carbon atoms), hexane (6 carbon atoms), heptane (7 carbon atoms), octane (8 carbon atoms), nonane (9 carbon atoms) and decane (10 carbon atoms).

Try to name the following compounds:

1. $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_3$
2. $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_3$
3. $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_3$

Try to draw structures for the following compounds:

4. propane
5. pentane
6. octane

II. The parent chain is numbered to give substituents the lowest possible numbers

Substituent names are methyl, ethyl, propyl, butyl, etc. The number showing the point of attachment to the parent chain precedes the substituent name. If you have more than one substituent with the same name, a number of attachment must be given for each substituent and the number of the substituents must be designated with di-, tri-, etc.

Try to name the following compounds...



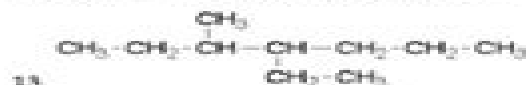
Try to draw structures for the following compounds...

10. 2,2-dimethylpropane
11. 3-methylheptane
12. 4,5-diethylnonane

III. Substituents are named in alphabetical order

Remember that the numbering of the parent chain must give all substituents the lowest possible numbers regardless of their names.

Try to name the following compound...



IV. A large substituent is numbered to give its point of attachment

The point of attachment is number one and any other smaller groups are named as substituent groups on the larger group. This numbering is independent of the numbering of the parent chain. Try to name the following compounds...

- 14.

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Marcel A. Müller



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