

2. Based on the Type of Bond

a. Saturated bonds, if all the carbon bonds are single bonds ($-C-C-$).

Example:



b. An unsaturated bond, if it contains double ($-C=C-$) or triple ($-C\equiv C-$) bonds in the carbon-carbon bond. It is said to be unsaturated because the double bonds, both double and triple, can still experience bond breaking.

Example:

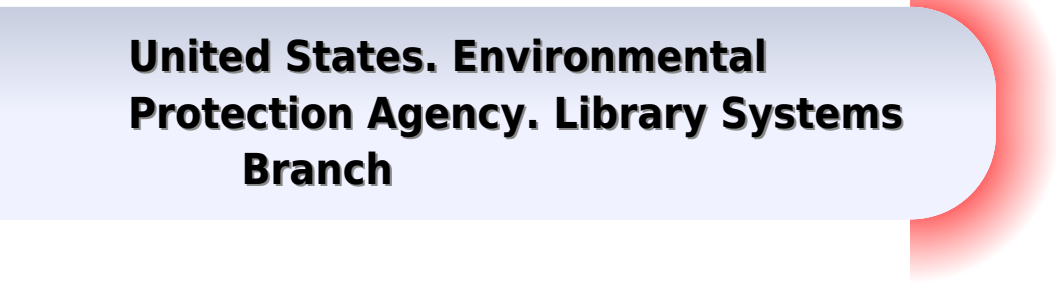


Exercise.

1. Explain the difference between organic and inorganic compounds!
2. How to test for the presence of the elements carbon, hydrogen, and oxygen in carbon compounds!
3. Explain the uniqueness of the carbon atom!
4. Explain the difference between aliphatic carbon chains and cyclic carbon chains along with examples of each!
5. Explain the difference between saturated bonds and unsaturated bonds in hydrocarbon compounds along with examples of each!

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Enjoying the Track of Expression: An Emotional Symphony within **Hydrocarbon Compounds Hydrocarb Planning Guide**

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