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## Ionic Compounds: Names and Formulas Worksheet

(a) magn	esium oxide	Mg0	(k) copper(i) bron	
(b) sodium fluoride		NaP	(I) tin(II) iodide	Sn Brz
(c) aluminum nitride		AIN	(m) iron(III) chlori	deFECI3
(d) potassium sulfide		K25	(n) calcium phosp	hide _ Caz Pz
(e) lithium iodide		LiT	(o) lead(II) cxide	Pbo
(f) calcium bromide		Cabra	(p) lead(IV) fluorid	e PbF4
(g) beryllium oxide		BeO	(q) tin(IV) bromide	Sn Bry
(h) nickel chloride		Nio	(r) copper(II) sulfid	1 -
(i) magnesium nitride		MgzNz	(s) iron(II) oxide	FeO
(i) aluminum sulfide			o sere and an excellent and a	
Write the r	names for the f	AlsS3	(t) calcium nitride	Ca3 N2
Write the r	names for the f	ollowing compounds.	(k) PbS	ead (II) su(ficle
Write the r e) Li <sub>2</sub> O b) AICl <sub>3</sub>	lithium aluminu	ollowing compounds. oxide un chloride	(k) PbS	ead (II) su(ficle in (IV) oxide
Write the r e) Li <sub>2</sub> O b) AICl <sub>3</sub>	lithium aluminu Magnes	ollowing compounds. oxide un chloride ium sulfide	(k) PbS(c) SnO <sub>2</sub> +	ead (II) su(ficle in (IV) oxide dium sulficle
Write the r a) Li <sub>2</sub> O b) AICI <sub>3</sub> c) MgS	lithium aluminu magnes	ollowing compounds. oxide un chloride ium sulfide m oxide	(k) PbS(c) (l) SnO <sub>2</sub> + (m) Na <sub>2</sub> S	ead (II) su(ficle in (II) oxide dium sulficle agnosium phosphic
Write the r a) Li <sub>2</sub> O b) AICl <sub>3</sub> c) MgS d) CaO	names for the f lithium aluminu magnes calcius	ollowing compounds. oxide  un chloride ium sulfide m oxide un bromide	(k) PbS	ead (II) su(ficle in (IV) oxide dium sulficle
Write the r e) Li <sub>2</sub> O b) AICl <sub>3</sub> c) MgS d) CaO	names for the f lithium aluminu magnes calcius	ollowing compounds. oxide un chloride ium sulfide m oxide	(k) PbS(c) (l) SnO <sub>2</sub>	ead (II) su(ficle in (II) oxide dium sulficle agnosium phosphic
	mames for the f lithium aluminu magnes calcius potassi beryllic	ollowing compounds. oxide  un chloride ium sulfide m oxide un bromide	(k) PbS	ead (II) su(ficle in (ID) oxide fium sulficle agresium phosphic ckel (II) oxide
Write the r (a) Li <sub>2</sub> O (b) AICI <sub>3</sub> (c) MgS (d) CaO (e) KBr	magras calcius potassi beryllic	ollowing compounds. oxide  un chloride ium sulfide m oxide un bromide um fluoride	(k) PbS	ead (II) su(ficle in (IV) oxide dium sulficle agnosium phosphic ckel (II) oxide opper (I) iodide
Write the r a) Li <sub>2</sub> O b) AICl <sub>3</sub> c) MgS d) CaO e) KBr	magnes calcius pofassi benyllic sodius	ollowing compounds. oxide un chloride ium sulfide m oxide un bromide um fluoride n ni fride	(k) PbS	ead (II) sulficle in (IV) oxide dium sulficle agnosium phosphic ckel (II) oxide opper (I) iodicle ad (IB) chloricle

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**K Morrison** 

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