

COMMON TYPES OF CHEMICAL REACTIONS

SYNTHESIS

Multiple reactants, which can be simple elements or compounds, combine together to form a single compound.



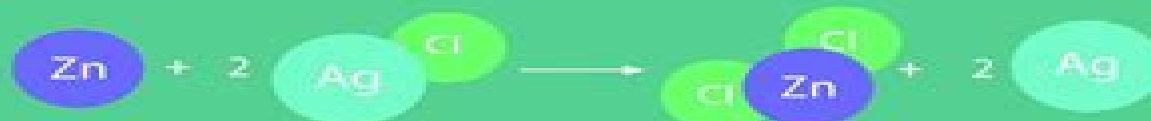
DECOMPOSITION

A compound breaks down into two or more simpler substances. Decomposition reactions are classified into thermal, electrolytic, and photo.



SINGLE-REPLACEMENT

One element is substituted for another element in a compound, generating a new compound and a pure element.



DOUBLE-REPLACEMENT

Two ionic compounds exchange cations or anions to form two new compounds. Forming a precipitate can help drive the reaction to the right.



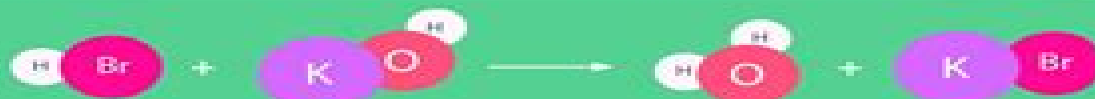
COMBUSTION

A fuel, typically a hydrocarbon, reacts with oxygen gas to form carbon dioxide and water, which generates heat and light.



NEUTRALIZATION

Acid and base combine to neutralize each other in aqueous solutions, usually resulting in a salt and water.



3 Types Of Chemical Reactions

CO Houle



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3 Types Of Chemical Reactions Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**3 Types Of Chemical Reactions**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve in to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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