

Time: 1.30 Hrs.

Standard 11 CHEMISTRY

Marks:

Part - I

Note: I) Answer

10×1=

II) Choose

Four alternative:

and write the option code and the corresponding answer.

1) Which of the following has highest hydration energy?

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2) Among the following aqueous solution, which has lowest boiling point

a) 0.1M C_2H_5O

b) 0.1M KCl

Original Question Paper

3) 5.85g of Sodium Chloride is dissolved in water and the solution was made up to 500ml using standard flask. The strength of the solution in molarity

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