

Gas Variables

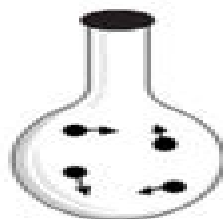
How are the variables that describe a gas related?

Why?

Imagine buying a balloon bouquet at a party store. How will the helium gas in the bouquet behave if you carry it outside on a hot summer day? How will it behave if you carry it outside during a snowstorm? What happens if the balloons are made of latex, which can stretch? What happens if the balloons are made of Mylar[®], which cannot stretch? What if you add just a small amount of gas to each balloon? What if you add a lot of gas? In this activity, you will explore four variables that quantify gases—pressure (P), volume (V), temperature (T), and moles (n) of gas. These four variables can be related mathematically so that predictions about gas behavior can be made.

Model 1 – Gases in a Nonflexible Container

Experiment A (Adding more gas)



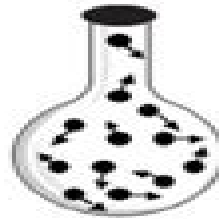
A1

Volume = 1 unit
External pressure = 1 atm
Internal pressure = 1 atm
Temperature = 200 K



A2

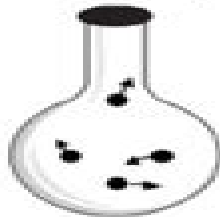
Volume = 1 unit
External pressure = 1 atm
Internal pressure = 2 atm
Temperature = 200 K



A3

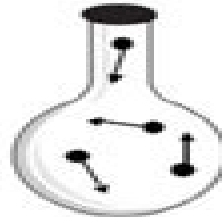
Volume = 1 unit
External pressure = 1 atm
Internal pressure = 3 atm
Temperature = 200 K

Experiment B (Heating the gas)



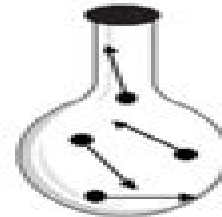
B1

Volume = 1 unit
External pressure = 1 atm
Internal pressure = 1 atm
Temperature = 200 K



B2

Volume = 1 unit
External pressure = 1 atm
Internal pressure = 2 atm
Temperature = 400 K



B3

Volume = 1 unit
External pressure = 1 atm
Internal pressure = 3 atm
Temperature = 600 K

*Note: Volume in this model is recorded in units rather than liters because 4 molecules of gas at the conditions given would occupy a very small space ($\sim 1 \times 10^{-22}$ mL). The particles shown here are much larger compared to the space between them than actual gas particles.

25 Gas Variables Packet Answers

**Barron's Educational Series, Brian
Stewart**



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Robert L. Evans, 2013-06-29 This book contains the proceedings of the International Symposium on Alternative and Advanced Automotive Engines held in Vancouver B C on August 11 and 12 1986 The symposium was sponsored by EXPO 86 and The University of British Columbia and was part of the specialized periods program of EXPO 86 the 1986 world's fair held in Vancouver Some 80 attendees were drawn from 11 countries representing the academic auto motive and large engine communities The purpose of the symposium was to provide a critical review of the major alternatives to the internal combustion engine The scope of the symposium was limited to consideration of combustion engines so that electric power for example was not considered This was not a reflection on the possible contribution which electric propulsion may make in the future but rather an attempt to focus the proceedings more sharply than if all possible propulsion systems had been considered In this way all of the contributors were able to participate in the sometimes lively discussion sessions following the presentation of each paper

Scientific and Technical Aerospace Reports, 1994 Lists citations with abstracts for aerospace related reports obtained from world wide sources and announces documents that have recently been entered into the NASA Scientific and Technical Information Database

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G. Davenport, Matthew J. King, 2005-11-04 More sulfuric acid is produced every year than any other chemical. It has a wide range of uses including phosphate fertilizer production, explosives, glue, wood preservatives, and lead acid batteries. It is also a particularly corrosive and dangerous acid with extreme environmental and health hazards if not manufactured, used, and regulated properly. *Sulfuric Acid Manufacture: Analysis, Control, and Optimization* keeps the important topics of safety and regulation at the forefront as it overviews and analyzes the process of sulfuric acid manufacture. The first nine chapters focus on the chemical plant processes involved in industrial acidmaking with considerable data input from the authors' industrial colleagues. The last 15 chapters are dedicated to the mathematical analysis of acidmaking. Both authors bring years of hands-on knowledge and experience to the work, making it an exceptional reference for anyone involved in sulfuric acid research and/or manufacture. Only book to examine the processes of sulfuric acid manufacture from an industrial plant standpoint as well as mathematical. Draws on the industrial connections of the authors through their years of hands-on experience in sulfuric acid manufacture. A considerable amount of industrial plant data is presented to support the text.

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