

Balancing Equations Worksheet II

Count the number of atoms, then use coefficients to balance each equation.

- 1) ____ AlBr_3 + ____ K ? ____ KBr + ____ Al
- 2) ____ FeO + ____ PdF_2 ? ____ FeF_2 + ____ PdO
- 3) ____ P_4 + ____ Br_2 ? ____ PBr_3
- 4) ____ LiCl + ____ Br_2 ? ____ LiBr + ____ Cl_2
- 5) ____ PbBr_2 + ____ HCl ? ____ HBr + ____ PbCl_2
- 6) ____ CoBr_3 + ____ CaSO_4 ? ____ CaBr_2 + ____ $\text{Co}_2(\text{SO}_4)_3$
- 7) ____ Na_3P + ____ CaF_2 ? ____ NaF + ____ Ca_3P_2
- 8) ____ Mn + ____ HI ? ____ H_2 + ____ MnI_3
- 9) ____ Li_3PO_4 + ____ NaBr ? ____ Na_3PO_4 + ____ LiBr
- 10) ____ CaF_2 + ____ Li_2SO_4 ? ____ CaSO_4 + ____ LiF
- 11) ____ HBr + ____ $\text{Mg}(\text{OH})_2$? ____ MgBr_2 + ____ H_2O
- 12) ____ LiNO_3 + ____ CaBr_2 ? ____ $\text{Ca}(\text{NO}_3)_2$ + ____ LiBr
- 13) ____ AgNO_3 + ____ Li ? ____ LiNO_3 + ____ Ag
- 14) ____ $\text{Si}(\text{OH})_4$ + ____ NaBr ? ____ SiBr_4 + ____ NaOH
- 15) ____ NaCN + ____ CuCO_3 ? ____ Na_2CO_3 + ____ $\text{Cu}(\text{CN})_2$

Challenge: ____ C_2H_6 + ____ O_2 ? ____ CO_2 + ____ H_2O

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Resources in Education, 1991

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