Airbag Lab (gas stoichiometry)

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Materials:

- Vinegar (5% acetic acid)
- Baking soda (sodium bicarbonate)
- Electronic scale

Graduated cylinders

- Ziploc bag
- Calculator

Procedure:

1. Balance the chemical reaction of baking soda and vinegar:

- 2. Get a Ziploc bag
- 3. Record:

temperature: ______ in the room.

- Determine the volume of your Ziploc bag in liters:
- 5. Calculate the mass of baking soda (sodium bicarbonate) required to completely fill your airbag with CO2.

 Calculate the volume vinegar (5% acetic acid) required to completely fill your airbag. Vinegar has an acetic acid, HC₂H₃O₂, concentration of 0.833 mol / L.

Airbag Stoichiometry Lab Work And Answers

Marco Cascella

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