

# Matter Practice Assessment

Name: \_\_\_\_\_

**SKILL** Label each of the following as an atom, element, or compound



**SKILL** Classify the following as homogeneous or heterogeneous mixtures:

a. jar of different colored jelly beans

**Heterogeneous**

c. margarine

**Homogeneous**

**SKILL** What happens to the properties of elements, when the elements are chemically combined in a molecule?

**The properties of the elements change once they are chemically combined together in a molecule.**

**SKILL** A solution of a mysterious chemical is bright aqua color. If aluminum foil is added to the mysterious chemical solution, the aqua color disappears, a strange odor is released, and the mixture takes on a bright blue color, and gas and a red solid forms.

a. What is an example of a physical property discussed?

**Answers can vary - could include: bright aqua color, odor, bright blue color, color of solid is red.**

b. What is an indication that a chemical change has occurred?

**Answers can vary - could include: if foil is added the aqua color disappears, a strange odor is released after foil is added, mixture changes color, gas and solid form after mix of substances.**

**SKILL** Below is a description of a mixture that you are required to separate into its pure substance components. After considering the "ingredients" in the mixture outline what separation methods you would use, and in what order.

**Mix contains: Water, kool-aid, glass beads, and benzene (a liquid that does not mix with water).**

**Answer could vary, however the separation should be based on properties of components of the mixture.**

**E.g. 1) Freeze the water, kool-aid, glass bead and benzene mix until the benzene freezes and can be removed from remaining mix.**

**2) Pour the water, kool-aid and glass bead mixture through a filter. This will catch the glass beads, since they did not dissolve in water.**

**3) Either raise the temperature to boiling or allow the water to evaporate naturally. This will separate the water from the kool-aid, that will be left behind in the bottom of the container after the water evaporates.**

# Assessment Pretest Matter Answer Key

**M Mosston**



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