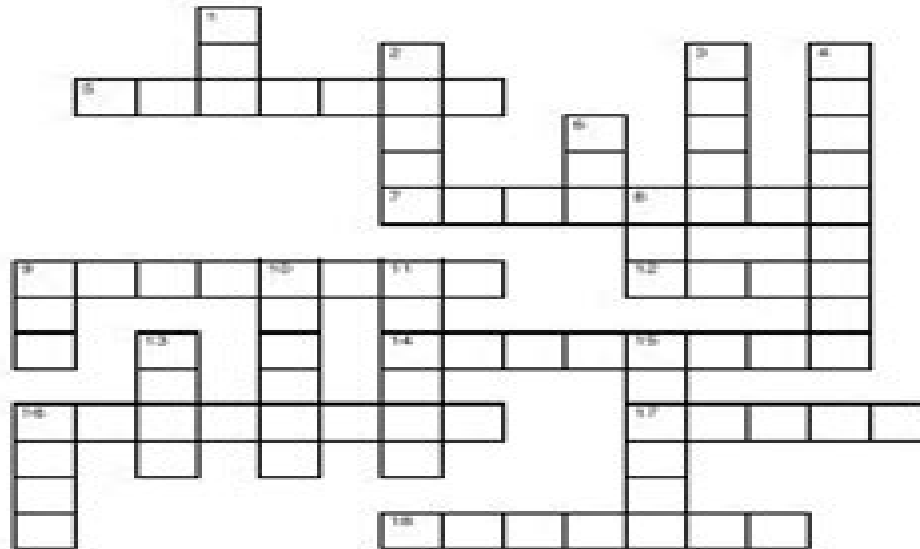


## Two Step Equation Crossword Puzzle



### Across

5.  $2y - 10 = 22$

7.  $2g - 3 = 35$

9.  $1 + 3z = 40$

12.  $7h + 2 = 65$

14.  $2v - 7 = 29$

16.  $2r - 6 = 22$

17.  $3g + 7 = 31$

18.  $3f - 7 = 38$

### Down

1.  $2m - 3 = 9$

2.  $4 + 3p = 25$

3.  $5k + 6 = 21$

4.  $2w - 8 = 26$

6.  $9 + 7d = 16$

8.  $4j - 8 = 32$

9.  $8k - 5 = 11$

10.  $4m - 23 = 57$

11.  $5b - 12 = 43$

13.  $7f + 6 = 34$

15.  $4 + 5m = 64$

16.  $9w - 8 = 37$

# 2 Step Equation Crossword Puzzle

**Hussin A.Rothana**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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web sep 24 2021 a diatomic molecule that consists of a polar covalent bond such as ce hf is a polar molecule as mentioned in section 4 7 because the electrons in the bond are nearer to the f atom this side of the molecule takes on a partial negative charge which is represented by  $\delta^-$   $\delta^-$  is the lowercase greek letter delta

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web polar bonding is a type of covalent bonding where the bonding electrons are distributed unequally between the bonding atoms pauling electronegativity values can be used to understand why some atoms form simple molecular compounds while other atoms bond together and form giant ionic lattices

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web q1 arrange the bonds in each of the following sets in order of increasing polarity c f be f o f o cl s br c p c s b f n o q2 determine based on electronegativity and symmetry if the following molecules are polar or nonpolar hcn h c n cf4 c f 4 cocl2 c o c l 2 nf3 n f 3 q3

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web when the atoms in a bond are not the same the bonding electrons are shared and the bond is electronegativities equally unequally non polar polar unequally polar the degree of polarity of a bond between any two atoms is determined by consulting a table of

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as bf<sub>3</sub> trigonal planar ch<sub>4</sub> tetrahedral pf<sub>5</sub> trigonal bipyramidal and sf<sub>6</sub>

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