

**20.1****DESCRIBING ACIDS AND BASES****SECTION REVIEW****Objectives**

- List the properties of acids and bases.
- Name an acid or base when given the formula.

**Key Terms**

- acid
- base

**Part A Completion**

After this completion exercise is done, your understanding of the concepts and terms is clear and reinforced in this section. Each blank has the compound with a name, about phrase, or a symbol.

Aqueous solutions of \_\_\_\_\_ 1 \_\_\_\_\_ are electrolytes. Acids react with many \_\_\_\_\_ 2 \_\_\_\_\_ to produce \_\_\_\_\_ 3 \_\_\_\_\_ gas. \_\_\_\_\_ 4 \_\_\_\_\_ are compounds that react with acids to form \_\_\_\_\_ 5 \_\_\_\_\_ and a salt.

Acids produce \_\_\_\_\_ 6 \_\_\_\_\_ ions when dissolved in water. There is a standard system for naming acids. For example,  $\text{HBr}$  is called \_\_\_\_\_ 7 \_\_\_\_\_ acid, while  $\text{HNO}_3$  is called \_\_\_\_\_ 8 \_\_\_\_\_ acid.

Bases produce \_\_\_\_\_ 9 \_\_\_\_\_ ions when dissolved in water. Some compounds that are \_\_\_\_\_ 10 \_\_\_\_\_ are named in the same way as other ionic compounds.

1. \_\_\_\_\_

2. \_\_\_\_\_

3. \_\_\_\_\_

4. \_\_\_\_\_

5. \_\_\_\_\_

6. \_\_\_\_\_

7. \_\_\_\_\_

8. \_\_\_\_\_

9. \_\_\_\_\_

10. \_\_\_\_\_

**Part B True-False**

Circle the word of these classes as true or false: true, AC; not true or false, NCF; or never true, NT.

- \_\_\_\_\_ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- \_\_\_\_\_ 12. An acid contains one or more hydroxide ions.
- \_\_\_\_\_ 13. Chemists have a special system for naming bases.

## 20 Describing Acids And Bases Section Review

**William H. Brown, Thomas Poon**



## 20 Describing Acids And Bases Section Review:

**Quantitative Chemical Analysis** Daniel C. Harris, 2010-04-30 QCA is the bestselling textbook of choice for analytical chemistry. It offers a modern portrait of the techniques of chemical analysis backed by a wealth of real world applications. This edition features new coverage of spectroscopy and statistics, new pedagogy and enhanced lecturer support. *Organic Chemistry Study Guide with Solutions Manual* Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material, chapter outlines, detailed comments for each chapter section, a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer. *Veterinary Anesthesia and Analgesia* Kurt A. Grimm, Leigh A. Lamont, William J. Tranquilli, Stephen A. Greene, Sheila A. Robertson, 2015-05-11 *Veterinary Anesthesia and Analgesia* the Fifth Edition of Lumb and Jones is a reorganized and updated edition of the gold standard reference for anesthesia and pain management in veterinary patients. Provides a thoroughly updated edition of this comprehensive reference on veterinary anesthesia and analgesia combining state of the art scientific knowledge and clinically relevant information. Covers immobilization, sedation, anesthesia and analgesia of companion, wild, zoo and laboratory animals. Takes a body systems approach for easier reference to information about anesthetizing patients with existing conditions. Adds 10 completely new chapters with in depth discussions of perioperative heat balance, coagulation disorders, pacemaker implantation, cardiac output measurement, cardiopulmonary bypass, shelter anesthesia and pain management, anesthetic risk assessment, principles of anesthetic pharmacology and more. Now printed in color with more than 400 images. Study Guide and Solutions Manual Neil E. Schore, 2002-08-02 **Quantitative Chemical Analysis, Sixth Edition** Daniel C. Harris, 2003 For instructors who wish to focus on practical industrial or research chemistry. Includes case studies, applications boxes and spreadsheet applications. *Physical Chemistry and Its Biological Applications* Wallace Brey, 2012-12-02 *Physical Chemistry and Its Biological Applications* presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being applied to increase understanding of living systems. Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes. Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes. Chapters 6 to 8 look at acid base equilibria and the link between electromagnetic radiation and the structure of atoms. Chapters 9 to 11 cover different types of bonding, the rates of chemical reactions and the process of adsorption. Chapters 12 to 14 present molecular aggregates, magnetic resonance spectroscopy and photochemistry and radiation. This book is useful to biological scientists for self study and reference. With modest additions of mathematical material by the teacher, the book should also be suitable for a full year major's course in physical chemistry. Understanding the Cell Denson K. McLain, 1996-01-16 *Understanding the Cell* is a comprehensive text at the college introductory level written in an easy to read conversational format. Within each section, key words are introduced, emboldened, discussed, then reviewed prior to moving on to the next subject. The key concepts are also illustrated. In addition, more than

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**Introduction to Organic Chemistry** William H. Brown,Thomas Poon,2016-01-13

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Baumle, 2018-08-31 Designed for the Diploma of Nursing Foundations of Nursing Enrolled Nurses Australia and New Zealand edition is mapped to the HLT54115 training package competencies and aligns to the revised Standards for Practice for the Enrolled Nurse Written to equip the enrolled nurse with current knowledge and basic problem solving and critical thinking skills to successfully meet the demanding challenges of today's health care the text clearly explains concepts and definitions and scaffolds knowledge The student friendly text provides a clear and fresh approach to the study of nursing it is straightforward and heavily illustrated with colour photos of procedures

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