CHAPTER 19 REVIEW

Oxidation-Reduction Reactions

SECTION 1

SHORT ANSWER Answer the following questions in the space provided.

- All the following equations involve redex reactions except
 - (a) CuO + H₂O → Ca(OH)₂.
 - (b) 2SO₂ + O₂ → 2SO₃.
 - (c) $2HgO \rightarrow 2Hg + O_2$.
 - (d) SnCl₄ + 2FeCl₂ → 2FeCl₃ + SnCl₂.
- Assign the correct exidation number to the individual atom or ion below.
 - _____ a. Ma in MaO₂
 - _____ b. S in S_R
 - e. Cl in CaCl₂
 - ______ d. 1 is 10₃
 - ______ e. C in HCO₃
 - f. Fe in Fe₂(SO₄)₃
 - _____ g. S in Fe₂(SO₄)₃
- 3. In each of the following half-reactions, determine the value of x.
 - _____ a. $S^{6+} + x e^- \rightarrow S^{2-}$
 - _____ b. $2Br'' \rightarrow Br_2 + 2e^-$
 - _____ c. $Sn^{4+} + 2e^- \rightarrow Sn^a$
 - d. Which of the above half-reactions represent reduction processes?
- 4. Give examples, other than those listed in Table 1 on page 631 of the text, for the following:
 - ______ n. a compound containing H in a −1 oxidation state
 - ______ b. a peroxide
 - e. a polyatomic son is which the oxidation number for S is +4
 - d. a substance in which the excidation number for F is not −1

20 Chapter 20 Oxidation Reduction Reactions Worksheet

KJ Lindholm-Leary

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Chemistry Workbook For Dummies Peter J. Mikulecky, Katherine Brutlag, Michelle Rose Gilman, Brian Peterson, 2008-08-06 From liquids and solids to acids and bases work chemistry equations and use formulas with ease Got a grasp on the chemistry terms and concepts you need to know but get lost halfway through a problem or worse yet not know where to begin Have no fear this hands on guide helps you solve many types of chemistry problems in a focused step by step manner With problem solving shortcuts and lots of practice exercises you ll build your chemistry skills and improve your performance both in and out of the science lab You ll see how to work with numbers atoms and elements make and remake compounds understand changes in terms of energy make sense of organic chemistry and more 100s of Problems Know where to begin and how to solve the most common chemistry problems Step by step answer sets clearly identify where you went wrong or right with a problem Understand the key exceptions to chemistry rules Use chemistry in practical applications with confidence **AP Chemistry For Dummies** Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out or your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam **Class 9 Chemistry Quiz**

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