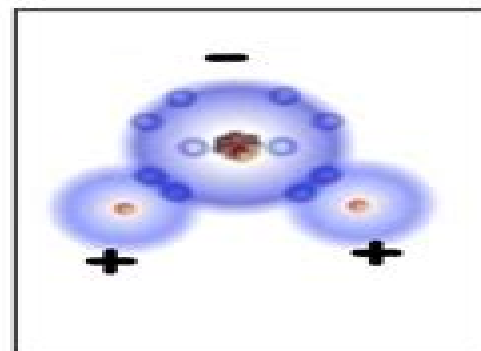
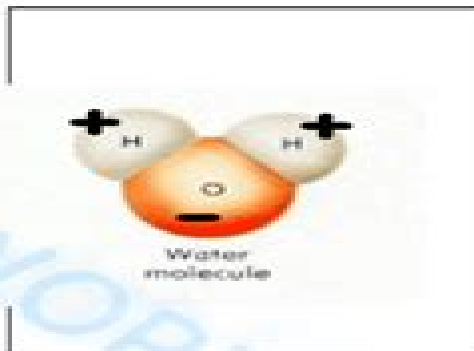
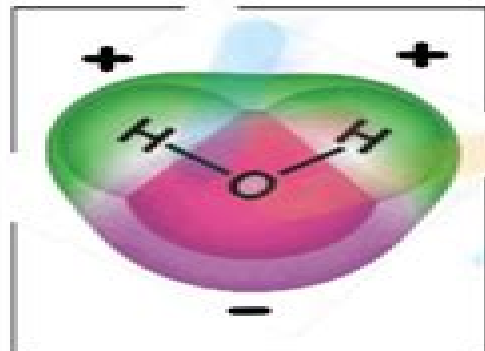


Properties of Water Activity

Name _____ Class _____

Water is everywhere. It's in the air we breathe. It's in our sink faucets. It's in every cell of our body. Water is an unusual substance with **many special properties**.

INTRODUCTION: The Water Molecule



What do you observe about the water molecule in these models?

1. _____
2. _____
3. _____

READ: Water is a polar molecule. That is, while a water molecule is neutral as a whole, it has a slightly positive charge on the oxygen side and a slightly negative charge on the hydrogen side. Water is found on Earth as a solid, liquid, and gas. This is because Earth is a very special planet with just the right range of temperatures and air pressures. Earth is said to be at the **triple point** for water.

QUESTIONS:

1. How many atoms are in a water molecule? Be specific.
2. Why does a water molecule have **polarity**?
3. Why is Earth said to be at the **triple point** for water?

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Fiona Baxter,Liz Dilley,2014-05-22 Cambridge Primary Science is a flexible engaging course written specifically for the Cambridge Primary Science curriculum framework This Teacher s Resource for Stage 5 contains guidance on all components in the series Select activities and exercises to suit your teaching style and your learners abilities from the wide range of ideas presented Guidance includes suggestions for differentiation and assessment and supplementing your teaching with resources available online to help tailor your scheme of work according to your needs Answers to questions from the Learner s Book and Activity Book are also included The material is presented in editable format on CD ROM as well as in print to give you the opportunity to adapt it to your needs *Green Technologies: Concepts, Methodologies, Tools and Applications*
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