

# 2.1

## CHEMICAL REACTIONS

### Section Review

#### Objectives

1. Measure enthalpy using a calorimeter.
2. Identify and predict chemical reactions using phase diagrams.
3. Apply the law of conservation of energy to chemical reactions.

#### Vocabulary

- |                  |               |                                  |
|------------------|---------------|----------------------------------|
| 1. endothermic   | 2. exothermic | 3. enthalpy                      |
| 4. heat capacity | 5. joule      | 6. law of conservation of energy |

### Part A Completion

Write the number(s) that apply to each of the following (check all that apply, and write none if none apply):

- |  |   |
|--|---|
| _____ 1. Enthalpy change depends on the direction of the reaction. | 1. <input checked="" type="checkbox"/> exothermic   |
| _____ 2. Enthalpy is a state function.                             | 2. <input checked="" type="checkbox"/> endothermic  |
| _____ 3. Enthalpy is a state function.                             | 3. <input checked="" type="checkbox"/> exothermic   |
| _____ 4. Enthalpy is a state function.                             | 4. <input checked="" type="checkbox"/> endothermic  |
| _____ 5. Enthalpy is a state function.                             | 5. <input checked="" type="checkbox"/> exothermic   |
| _____ 6. Enthalpy is a state function.                             | 6. <input checked="" type="checkbox"/> endothermic  |
| _____ 7. Enthalpy is a state function.                             | 7. <input checked="" type="checkbox"/> exothermic   |
| _____ 8. Enthalpy is a state function.                             | 8. <input checked="" type="checkbox"/> endothermic  |
| _____ 9. Enthalpy is a state function.                             | 9. <input checked="" type="checkbox"/> exothermic   |
| _____ 10. Enthalpy is a state function.                            | 10. <input checked="" type="checkbox"/> endothermic |
| _____ 11. Enthalpy is a state function.                            | 11. <input checked="" type="checkbox"/> exothermic  |
| _____ 12. Enthalpy is a state function.                            | 12. <input checked="" type="checkbox"/> endothermic |
| _____ 13. Enthalpy is a state function.                            | 13. <input checked="" type="checkbox"/> exothermic  |
| _____ 14. Enthalpy is a state function.                            | 14. <input checked="" type="checkbox"/> endothermic |
| _____ 15. Enthalpy is a state function.                            | 15. <input checked="" type="checkbox"/> exothermic  |
| _____ 16. Enthalpy is a state function.                            | 16. <input checked="" type="checkbox"/> endothermic |
| _____ 17. Enthalpy is a state function.                            | 17. <input checked="" type="checkbox"/> exothermic  |
| _____ 18. Enthalpy is a state function.                            | 18. <input checked="" type="checkbox"/> endothermic |
| _____ 19. Enthalpy is a state function.                            | 19. <input checked="" type="checkbox"/> exothermic  |
| _____ 20. Enthalpy is a state function.                            | 20. <input checked="" type="checkbox"/> endothermic |

### Part B True-False

Write true or false for each statement. If the statement is true, write true. If the statement is false, write false.

1. Enthalpy is a state function. ☒ True
2. Enthalpy is a state function. ☒ True
3. Enthalpy is a state function. ☒ True
4. Enthalpy is a state function. ☒ True
5. Enthalpy is a state function. ☒ True
6. Enthalpy is a state function. ☒ True
7. Enthalpy is a state function. ☒ True
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13. Enthalpy is a state function. ☒ True
14. Enthalpy is a state function. ☒ True
15. Enthalpy is a state function. ☒ True
16. Enthalpy is a state function. ☒ True
17. Enthalpy is a state function. ☒ True
18. Enthalpy is a state function. ☒ True
19. Enthalpy is a state function. ☒ True
20. Enthalpy is a state function. ☒ True

# 1describing Chemical Reactions Section Review

**Benjamin Lillard**



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system to self organize its processes and components c the increase in capacity of the system for learning and adaptation to  
external change d all of the above

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areas of rock b pollutants cling to the materials that make up the aquifer and contaminate the clean water c the recycling  
process of groundwater can take hundreds or thousands of years

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