

# 2.1

## CHEMICAL REACTIONS

### Section Review

#### Objectives

1. Measure enthalpy using a calorimeter.
2. Identify and predict chemical reactions using thermodynamic data.
3. Apply the law of conservation of energy to chemical reactions.

#### Vocabulary

- |                    |               |                       |
|--------------------|---------------|-----------------------|
| 1. endothermic     | 2. exothermic | 3. enthalpy           |
| 4. enthalpy change | 5. product    | 6. thermodynamic data |

### Part A: Completion

Use the information available to identify a thermodynamic state function, and write a balanced chemical equation for the reaction that takes place. Indicate whether the reaction is endothermic or exothermic.

- |  |           |
|--|-----------|
| 1. Enthalpy change decreases when a substance _____.             | 6. _____  |
| 2. Enthalpy change is a _____ consequence of a substance change. | 7. _____  |
| 3. Enthalpy change is a _____ change. It has a negative value.   | 8. _____  |
| 4. Enthalpy change is a _____ change. It has a positive value.   | 9. _____  |
| 5. _____ is the change in enthalpy for a reaction.               | 10. _____ |
| 6. _____ is the change in enthalpy for a reaction.               | 11. _____ |
| 7. Enthalpy change is a _____ change. It has a _____ value.      | 12. _____ |
| 8. Enthalpy change is a _____ change. It has a _____ value.      | 13. _____ |
| 9. Enthalpy change is a _____ change. It has a _____ value.      | 14. _____ |
| 10. Enthalpy change is a _____ change. It has a _____ value.     | 15. _____ |
| 11. Enthalpy change is a _____ change. It has a _____ value.     | 16. _____ |
| 12. Enthalpy change is a _____ change. It has a _____ value.     | 17. _____ |
| 13. Enthalpy change is a _____ change. It has a _____ value.     | 18. _____ |
| 14. Enthalpy change is a _____ change. It has a _____ value.     | 19. _____ |
| 15. Enthalpy change is a _____ change. It has a _____ value.     | 20. _____ |

### Part B: True-False

Indicate whether each statement is true or false. If false, explain why.

1. Enthalpy change is a state function.
2. Enthalpy change is a state function.
3. Enthalpy change is a state function.
4. Enthalpy change is a state function.
5. Enthalpy change is a state function.
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19. Enthalpy change is a state function.
20. Enthalpy change is a state function.

# 1describing Chemical Reactions Section Review

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**Effiong Eyo**

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