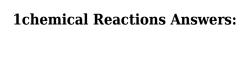
Excretion

CHEMICAL REACTIONS AND

	AUS	HONS
Objection	ve Questions	
Multiple Choice Questions	Questions	
1. Which of the		
1. Which of the statements about to (i) Lead is only 2PbO(s) + C	he reaction below	
(i) Lead is getting reduced	$(s) \longrightarrow 2Pb(S) + CO_2(2)$	
(a) Carbon dioxide in	22	
(iii) Carbon is getting oxidised	Tised	
The state of the s		
7. 1.7	(b) (i) and (iii)	
(c) (i), (ii) and (iii)	(d) all of these.	
2. Fe ₂ O ₃ + 2AI - Al ₂ O ₃ + 2Fe The above reaction is an example		(2924)
(a) combination reaction		
(c) decomposition reaction	(b) double displacement rea	ction
. What happens when dilute byde-	(d) displacement reaction.	
(a) Hydrogen gas and iron chlorid	le are produced	ings ? (2022)
() cinornie gas and fron hydroxi	de are produced	
(c) 140 reaction takes place		
(d) Iron salt and water are produce	od.	
In the reaction 2FeCl ₂ + Cl ₂	2FeCl ₃ Chlorine is a-	
(a) Reducing agent	(b) Oxidizing agent	
(c) Catalyst	(d) Inert medium.	
Rancidity can be prevented in foo		
(a) Mixing antioxidant	(b) Keeping in refrigerator	
c) Keeping away from light	(d) All of these.	
Which among the following is not		
a) Oxygen	(b) Cone. Sulphuric acid	
c) Chlorine	(d) Hydrogen.	
Vhich among the following gases		items fresh ?
Carbon dioxide and Oxygen	(b) Nitrogen and helium	
) Helium and Oxygen	(d) Nitrogen and Oxygen.	
hemically rust is—	Chile all side on the State of	
) Hydrated ferrous oxide	(b) Ferric oxide	
) Hydrated ferric oxide	(d) None of these.	
exothermic reaction in which	clucose combines with oxyger	a in the cens of
r body and provides energy-		
Respiration	(b) Digestion	
Excretion	(d) None of these.	

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