

## CHEMICAL REACTIONS AND EQUATIONS

### Objective Questions

#### Multiple Choice Questions

1. Which of the statements about the reaction below are incorrect ?  

$$2\text{PbO(s)} + \text{C(s)} \longrightarrow 2\text{Pb(l)} + \text{CO}_2\text{(g)}$$

- (i) Lead is getting reduced
- (ii) Carbon dioxide is getting oxidised
- (iii) Carbon is getting oxidised
- (iv) Lead oxide is getting reduced
- (a) (i) and (ii)
- (b) (i) and (iii)
- (c) (i), (ii) and (iii)
- (d) all of these.

2.  $\text{Fe}_2\text{O}_3 + 2\text{Al} \longrightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$

The above reaction is an example of a—

(2024)

- (a) combination reaction
- (b) double displacement reaction
- (c) decomposition reaction
- (d) displacement reaction.

3. What happens when dilute hydrochloric acid is added to iron fillings ? (2022)

- (a) Hydrogen gas and iron chloride are produced
- (b) Chlorine gas and iron hydroxide are produced
- (c) No reaction takes place
- (d) Iron salt and water are produced.

In the reaction  $2\text{FeCl}_2 + \text{Cl}_2 \longrightarrow 2\text{FeCl}_3$  Chlorine is a—

- (a) Reducing agent
- (b) Oxidizing agent
- (c) Catalyst
- (d) Inert medium.

Rancidity can be prevented in food items by—

- (a) Mixing antioxidant
- (b) Keeping in refrigerator
- (c) Keeping away from light
- (d) All of these.

Which among the following is not an oxidizing agent ?

- (a) Oxygen
- (b) Conc. Sulphuric acid
- (c) Chlorine
- (d) Hydrogen.

Which among the following gases are used to keep the oily food items fresh ?

- (a) Carbon dioxide and Oxygen
- (b) Nitrogen and helium
- (c) Helium and Oxygen
- (d) Nitrogen and Oxygen.

Chemically rust is—

- (a) Hydrated ferrous oxide
- (b) Ferric oxide
- (c) Hydrated ferric oxide
- (d) None of these.

An exothermic reaction in which glucose combines with oxygen in the cells of our body and provides energy—

- (a) Respiration
- (b) Digestion
- (c) Excretion
- (d) None of these.

# 1chemical Reactions Answers

**Donald A. McQuarrie, Stanley Gill**



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